Document heading: Research Article

Synthesis and Antimicrobial Properties of Some Compounds

Nevin Çankaya^{1*}, Sevda Kırbağ²

¹Department of Chemistry, Faculty of Science, University of Uşak, Uşak, Turkey ²Department of Biology, Faculty of Science, University of Fırat, Elazığ, Turkey

Received: 20-10-2018 / Revised: 23-11-2018 / Accepted: 10-12-2018

Abstract

N-cyclohexylacrylamide (NCA), N-cyclohexylmethacrylamide this (NCMA) nitrophenyl)acrylamide (4NPA) amide-derivative monomer, 2-(bis(cyanomethyl)amino)-2-oxoethyl methacrylate (CMA2OEM) acrylate-derivative monomer and 2-(4-methoxyphenylamino)-2-oxoethyl methacrylate (MPAEMA) monomer and its homopolymer, and the resin of MPAEMA with 2-acrylamido-2-methyl-1-propanesulfonic acid (AMPS) were synthesized. The antimicrobial activity of compounds were tested Escherichiacoli ATCC 66032, Staphylococcus aureus COWAN 1, Bacillusmegaterium DSM 32, Enterobacteraeregenes CCM 2531Candida tropicalis ATCC 13803.

Keywords: Monomer; Polymer; Antimicrobial activity.

Introduction

The polymers are classified according to the type of mer which they have: Homopolymer, polymers formed by the incorporation of a single monomer. Copolymer, a polymer formed by two or more monomers. When a compound passes from the monomer structure to the polymer structure, there are many variations in physical, chemical and biological properties with increasing molecular weight. A number of studies have been performed in our laboratories on the synthesis of meth/acrylamide and meth/acrylate monomer and their radical polymerization. In the literature, there are many copolymer and hydrogel synthesis with AMPS monomer[1-5]. Copolymers of AMPS with ethylene dimethacrylate are used in contact lenses, and AMPS-g-styrene gives self-reinforced hydrogels [5]. These studies show that the nature, as well as position of the substituent, had a large effect on antimicrobial properties.

*Correspondence

Nevin Cankava

Department of Chemistry, Faculty of Science, University of Uşak, Uşak, Turkey.

E- Mail: nevin.cankaya@usak.edu.tr nevincankaya@hotmail.com

Microorganism contaminationis important fornumerous industries, including but not limited to medical devices. healthcare products, water purification systems, hospital and dental equipment, food storage and packaging. One way to avoid the microbial contamination is to develop materials antimicrobial properties. Therefore, biocidal substances have attracted much attention in recent years [6]. We report in this manuscript the synthesis and characterization of monomer, as well as homopolymer and resin. These compounds were also tested for their antimicrobial properties against microorganisms such as Escherichia coli, Staphylococcus aureus, Bacillus megaterium, Enterobacter aeregenes, Candida tropicalis.

Experimental

Experimental Studies

The amide-monomer N-cyclohexylacrylamide (NCA) and N-cyclohexylacrylamide (NCMA) (Fig. 1) were obtained by reaction of cyclohexylamine with meth/acryloyl chloride [7-9].

$$\begin{array}{c|c}
CI & H_2C = CH \\
 & C = O \\
 & C = O \\
 & CH = CH_2
\end{array}$$

N-cyclohexylacrylamide (NCA)

$$\begin{array}{c|c}
 & CH_3 \\
 & H_2C = C \\
 & C = CH_2 \\
 & CH_3
\end{array}$$

N-cyclohexylmethacrylamide (NCMA)

Figure 1: Synthesis of NCA/NCMAmonomer

N-(4-nitrophenyl)acrylamide (4NPA) (Fig. 2) [10] and 2-(bis(cyanomethyl)amino)-2-oxoethyl methacrylate (CMA2OEM) (Fig. 3) [11] monomer, and 2-(4-methoxyphenylamino)-2-oxoethyl methacrylate

(MPAEMA) monomer [12, 13] and its homopolymer (Fig. 4, 5) were synthesized by a method adapted from the literature in the our lab. Synthesis schemes are given in Fig. 2-5.

$$O_2N$$
 $NH_2 + C=O$
 $CH=CH_2$
 $NH_2C=CH$
 $C=O$
 $NH_2C=O$
 $NH_2C=O$

(4NPA)

Figure 2: Synthesis of 4NPA monomer

Figure 3: Synthesis of CMA2OEM monomer

e-ISSN: 2590-3241, p-ISSN: 2590-325X

2-chloro-*N*-(4-methoxyphenyl)acetamide

(I) +
$$H_2C = C - C - ONa \longrightarrow H_2C = C - C - O - C - C - N - O - CH_3$$

sodium methacrylate

2-(4-metoksifenilamino)-2-oksoetil metakrilat (MPAEMA)

Figure 4:Synthesis of MPAEMA monomer

$$\begin{array}{c} CH_3 \\ H_2C = C \\ O = C$$

Figure 5: Synthesis of MPAEMA homopolymer

Chelating resin poly(2-(4-methoxyphenylamino)-2-oxoethyl methacrylate-co-divinylbenzene-co-2-acrylamido-2-methyl-1-propanesulfonic acid)

(MPAEMA-co-DVB-co-AMPS) was prepared by the procedure described in the literature (Fig. 6) [14].

Figure 6: Synthesis of MPAEMA-co-DVB-co-AMPS resin

Microbial Strains, Culture Media and Antimicrobial Screening

Escherichia coli ATCC 66032, Staphylococcus aureus COWAN 1, Bacillus megaterium DSM Enterobacter aeregenes CCM 2531Candida tropicalis ATCC 13803 used in the study were provided by the culture collection of the Microbiology Laboratory of University of Firat. Antimicrobial tests were carried out by disc diffusion method using 100 µL of suspension containing 10⁶ cells / mL of bacteria, 10⁴ cells / mL yeast as per McFarland standard, inoculated into Mueller Hinton Agar (Difco), Malt Extract Agar (Difco) and Sabouroud Dextrose Agar (Oxoid), respectively. The discs (6 mm diameter) were impregnated with 100 µg of the compound, placed on the inoculated Mueller Hinton Agar (Difco), Malt Extract Agar (Difco), respectively. Steril petri dishes (9cm diameter) were placed at 4 °C for 2h. Then, the inoculated plates were incubated at 37±0.1°C at 24 h for bacterial strains and also at 25±0.1°C at 72 h for yeasts. Antimicrobial activity was evaluated by measuring the zone of inhibition against the test microorganism.Streptomycin sulfate (10 mg/disc) and Nystatin (30 mg/disc) were used as standard antibiotic[15].

Results and discussion

The resulting inhibition zones on the plates were measured (mm). The data reported in Table 1.

The results were standardized against Streptomysinsulfat and Nystatin under the same conditions. The compounds showed selective antimicrobial activities. The results show that most of the compounds did not inhibit the growth of the test microorganisms MPAEMA-co-DVB-co-(except AMPS). While MPAEMA mono and homopolymer showed no biological activity, MPAEMA-co-DVB-co-AMPS showed bio activity. This may be due to the many electronegative atoms present in AMPS. In the literature, it is seen that many polymers are synthesized by using AMPS monomer. Among these, it was observed that, when looking at the polymers with AMPS, they showed biological activity [1-4]. The results we found in this study are consistent with the literature.

Table 1:Antimicrobial effects of monomer, homopolymer, and copolymer (diameter zones of inhibition, mm)

Compound	E.coli	S.aureus	B.megaterium	E. aeroginosa	C.tropicalis
NCA monomer	-	8	-	=	=
NCMA monomer	-	7	-	=	=
4NPA monomer	-	-	-	=	=
CMA2OEM	-	-	-	8	9
monomer					
MPAEMA mono	-	-	-	-	-
and					
homopolymer					
MPAEMA-co-	10	12	8	12	8
DVB-co-AMPS					
Standard	10**	9**	13**	15**	12*
antibiotics					

Compound concentration = 0.1 mg/disk; (-) the compounds have any activity against the microorganism. **: Nystatin (Antifungal, 30 µg/disc), *: Streptomysinsulfat (antibacterial, 10 µg/disc).

Conclusion

study, N-cyclohexylacrylamide, this cyclohexylmethacrylamide, N-(4nitrophenyl)acrylamide, 2-(bis(cyanomethyl)amino)-2oxoethyl methacrylate and 2-(4-methoxyphenylamino)-2-oxoethyl methacrylate (MPAEMA) monomer and its homopolymer, and the resin of MPAEMA with 2acrylamido-2-methyl-1-propanesulfonic acid (AMPS) (MPAEMA-co-DVB-co-AMPS) were synthesized in our laboratory. This compounds were tested for its antimicrobial activity against microorganism. Some of the compounds prevented the development of microorganisms while others did not affect. We believe that this study will guide the biological properties of the substances to be synthesized in the future.

References

- 1. C. Soykan, R. Coskun, S. Kirbag, Poly(crotonic acid-co-2-acrylamido-2-methyl-1-propane sulfonic acid)-metal complexes with copper(II), cobalt (II), and nicke l(II): Synthesis, characterization and antimicrobial activity, European Polymer Journal, 2007;43: 4028-
- 2. C. Soykan, R. Coskun, S. Kirbag, E. Şahin, Synthesis, Characterization and Antimicrobial of Poly(2-acrylamido-2-Activity methyl-1-propanesulfonicacid-co-crotonic acid), Journal of Macromolecular Science Part A: Pure and Applied Chemistry, 2007;44: 31–39.

- 3. C. Soykan, R. Coskun, A. Delibas, E. Şahin, Copolymers of 2-Acrylamido-2 Methyl-1-Propanesulfonic Acid/Maleic Acid: Synthesis, Characterization and Antimicrobial Activity, Chinese Journal of Polymer Science, 2007;25 (5):491-500.
- 4. A. Delibas, C. Soykan, Novel Copolymers of N-(4-Bromophenyl)-2-Methacrylamide 2-Acrylamido-2-Methyl-1-Propanesulfonic Acid Journal of Macromolecular Science, Part A: Pure and Applied Chemistry, 2007;44:969–975.
- 5. M. Matsukata, M. Hirata, J.P. Gong, Y. Osada, Y. Sakurai, T. Okano, Two-step surfactant binding of solvated and cross-linked poly(Nisopropylacrylamide-co-(2-acrylamido-2methyl propane sulfonic acid)), Colloid and Polymer Science, 1998;276 (1): 11-18.
- 6. C Soykan, Ş. Güven, R. Coşkun, Copolymers of 2-[(5-Methylisoxazol-3-yl)amino]-2-oxo-ethyl Methacrylate with Ethyl Methacrylate: Monomer Reactivity Ratios, Thermal Properties and Antimicrobial Activity, Journal Macromolecular Science, Part A:Pure and Applied Chemistry, 2006;43: 1619–1633.
- 7. N. Çankaya, M. M. Temüz, Monomer Reactivity Ratios of Cellulose Grafted with cyclohexylacrylamide and Methyl Methacrylate by Atom Transfer Radical Polymerization, Cellulose Chemistry and Technology 2014;48 (3-4), 209-215.

e-ISSN: 2590-3241, p-ISSN: 2590-325X

- **8.** F. Akman, N. Cankaya, A study of experimental and theoretical analysis of N-cyclohexylmethacrylamide monomer based on DFT and HF computations, Pigment & Resin Technology, 45 (5), 2016.
- M. Erdogan, Y. Acikbas, C. Soykan, N. Cankaya and R. Capan, Optical and chemical sensor properties of Langmuir-Blodgett thin films coated with N-cyclohexylmethacrylamide monomer, Journal of Optoelectronics and Advanced Materials, 2018;20:9-10.
- 10. N. Çankaya, M. M. Temüz, Characterization and Monomer Reactivity Ratios of Grafted Cellulose with N-(4-nitrophenyl)acrylamide and Methyl Methacrylate by Atom Transfer Radical Polymerization, Cellulose Chemistry and Technology, 2012;46 (9-10), 551-558.
- 11. E. B. Sas, N. Çankaya, M. Kurt, Synthesis of 2-(bis (cyanomethyl) amino)- 2- oxoethyl methacrylate monomer molecule and its characterization by experimental and theoretical methods, Journal of Molecular Structure, 2018; 1161: 433-441.

- **12.** N. Çankaya, G. Besci, Synthesis, characterization, thermal properties and reactivity ratios of methacrylate copolymers including methoxy group, Journal of the Faculty of Engineering and Architecture of Gazi University, 2018;33:3: 1155-1170.
- 13. Y. Acikbas, N. Cankaya, R. Capan, M. Erdogan, C. Soykan, Swelling behaviour of the 2-(4-methoxyphenylamino)-2-oxoethyl methacrylate monomer LB thin film exposed to various organic vapours by quartz crystal microbalance technique, Journal of Macromolecular Science, Part A, Pure and Applied Chemistry, 2016;53(1):18–25.
- 14. T. Daşbaşı, Ş. Saçmacı, N. Çankaya C. Soykan, A New Synthesis, Characterization and Application Chelating Resin for Determination of Some Trace Metals in Honey Samples by FAAS, Food Chemistry, 2016;203:283–291.
- **15.** C.M. Collins, P.M. Lyne, Microbiological methods, Butterworths-Heinemann, London, 1989, ISBN: 0 340 80896 9

Source of Support:Nil Conflict of Interest: Nil